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*Second Quarterly Report*

**STUDY OF PROCESS VARIABLES ASSOCIATED  
WITH MANUFACTURING HERMETICALLY-SEALED  
NICKEL-CADMIUM CELLS**

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GREENBELT, MARYLAND 20771



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## ABSTRACT

This report describes the second phase of experiments designed to determine the critical process variables in the manufacturing of aerospace nickel-cadmium cells. This second phase included additional analyses of plaque sintering experimental data and setting up of the impregnation and polarization process for cell plates. Also covered in this phase was a study of separator materials and ceramic-to-metal seals.

Work continued on the analysis of the plaque manufacturing data. Analysis of variance has shown that 85% of the total variation occurs within any individual plaque. It was further concluded that a 2% variation is introduced into the plaque by virtue of the manufacturing process.

Regression equations were determined with and without interactions. The results were compared and it was determined that there are significant interactions for this manufacturing process. Temperature, belt speed and dewpoint effects are non-linear due to these interactions. Hot zone temperature, belt speed, dewpoint, weight and second cooling zone temperature cause relatively strong effects on plaque characteristics. Belt speed is easily controlled but improved controls are recommended for dewpoint and temperatures of hot zones and cooling temperatures.

Additional use was made of the regression equation to predict variability of the plaque manufacturing process within the normal tolerance range of the equipment. The following conclusions were made as a result of this variability study for two (2) particular equipment settings.

- A. An increase of 30°F in temperature at 1760°F causes a 2% decrease in void and thickness and an increase in strength of 12%. However, this same increase at 1860°F can cause no change in void or thickness and can cause diminished strengths of 5-6%.
- B. An increase in belt speed of 1 inch/minute will cause less than a 1% change in void and thickness. Strength changes of less than 2% were shown at 1760°F but a 6% strength decrease was predicted at 1860°F with this increase in belt speed.
- C. Dewpoint exhibits especially strong affects for only a plus or minus change of 5°F. Thickness and void changes of 5% were seen at 1860°F with strength changes of up to 40%. Dewpoint effects are most pronounced on the lightest weight plaques.
- D. A change of 20 cfh of atmosphere amount causes no change in void or thickness and less than 1% in strength.
- E. Close spacing is most desirable for uniform plaques.
- F. A 16°F increase in temperature of the first cooling zone water increases thickness and void in the range of 0.0 to 0.7%, and increases strength approximately 2.5%.
- G. A 16°F increase in temperature of the second cooling zone water predicted large increases in strength and significant changes in void and thickness.
- H. A plaque weight increase will increase thickness and void. Strength is increased at 1760°F as plaque weight is increased, but at 1860°F strength is decreased as plaque weight is increased.

## I. INTRODUCTION

The objective of this program is to develop a process procedure and control for manufacturing nickel-cadmium aerospace cells with reliable five (5) year life capability. In order to achieve these objectives, each component part will be investigated separately and collectively to determine the critical variables and related interactions.

The total program consists of four (4) distinct, yet interrelated phases. The first phase consisted of a detailed analysis of our procedures in conjunction with a review of pertinent literature of nickel-cadmium batteries to assess critical variables of the various processes that affect cell performance. The second phase will involve the evaluation and testing (verification) of the variables and their interrelation as determined in Phase 1. This will include a design of experiments to experimentally identify critical variables and to establish tolerances required for uniform performance. Phase 3 includes the detailed preparation of a Quality and Reliability Assurance Program, Acceptance and Manufacturing Flow Sheets and a complete hardware specification that can be practically implemented in a cost effective manner. This specification will be patterned after "S-716-P-23, Interim Model Specification for High Reliability Nickel-Cadmium Spacecraft Cells." The Fourth Phase of the program will be to implement the results of Phases 1 through 3 on a production basis. This effort will "prove out" the conclusions and will establish both validity of concept and applicability to production equipment and overall operational capability. During this phase, the deliverable items of separation, positive and negative plates will be prepared. Also, 20 nickel-cadmium cells of 20 ampere-hour size will be manufactured to the developed procedure. Inspection levels will be 100% minimum and complete traceability maintained.

The first quarter of this program was devoted to investigating the dry-sintering process used in manufacturing porous nickel plaque. A factorial experiment was designed to examine the sintered plaque characteristics as a function of the process variables. The data gathered from this experiment were analyzed using a step-wise multiple regression technique designed for use with the IBM 1130 computer. At the completion of analysis, plaques with different characteristics were selected for use in the impregnation factorial experiment. After the impregnation study, these plaques will be characterized both electrically and physically to determine the effects of sintering and impregnation variations. Tolerance limits will be selected and plates will be produced for a production lot of cells. Studies are also being conducted on other component parts, such as, separators, ceramic-to-metal seals, welding techniques, etc. At the completion of these component studies, cells will be built and investigated for such things as electrolyte amount, positive-negative ratio, compression, etc. The cells produced at the completion of the program will be placed on Life Cycle Testing and cycled to failure.

## II. ANALYSIS OF DATA-PLAQUE MANUFACTURING STUDY

### A. Computer Programming

At the close of the last Interim Report, Run No. 9 (mid level run) had been completed but the data had not been included in the eight (8) previous plaque runs for analysis. The 8 previous run analysis had been used to predict the outcome of Run No. 9 and a series of comparisons (actual vs. predicted) were presented. Several changes were instituted to shorten the (future) time of analyzing the data by the Gardner Multiple Regression Program. A disk file was setup and the regression program modified to read this file. The file is loaded with up to 60 variables and 750 observations (total to date) using a separate scaling and transformation program. The scaling (-1 to +1) routine is not part of the Gardner Program (See Section B below). This change in procedure resulted in a significant reduction in computer running time.

### B. Reason For and Procedure For Coding of Independent Variables

The raw data consists of independent variables in the original engineering units; for example, temperatures of approximately 1700°F to thicknesses of 0.025", etc. The resulting coefficients would range from very small numbers to very large numbers. This presents difficulties as very large numbers or very small numbers present computation errors or numbers larger than the capability of the computer program. In order to "level" the values, a system is followed that reproducibly scales numbers so they reflect the actual run values and provides ready comparison among the coefficients. A program is available to "code" the variables by running through each variable to find the maximum and minimum value of each and then compute each variable to

find the maximum and minimum value of each and then compute each variable by proration giving a +1.000 value to the maximum and -1.000 to the minimum. The response or responses are, of course, not changed. The "raw" data is now "coded". The maxima and minima for each are punched in a new deck for future coding of prediction levels if desired. Rounding errors and subtraction and addition errors are lessened. The standard precision of the IBM 1130 is 6 figure accuracy so if numbers are subtracted larger than 6 figures, an error will be introduced. A discussion of these latter points is in Draper and Smith, Applied Regression Analysis, p. 148, John Wiley, 1966.

In order to improve the accuracy of the models, interactions of the variables are introduced as additional variables. To lower the number of variables, only first order interactions are used (e.g. temperature x dewpoint, etc). Fitting a curve (equation) is simplified by making the product as raw data and then coding the new variable. Thus, the product range from product of smaller numbers (-1) to larger numbers (+1). If the interactions were made from coded original variables, the progression of numbers would be more complex since a +1 could arise from the product of 2 large numbers (+1 x +1) as well as from 2 small numbers (-1 x -1). The computer program, thus builds the new variables (interactions) before the coding takes place. The data is built into a permanent disk file so from one raw data (original variables) deck, any new data set with interactions can be available for the regression programs.

### C. Error Variance Analysis

Analysis of variance for these data requires some modification from that usually performed on data. Each plaque resulted in 9 samples (observations) and each run produced 6 plaques. There is thus an error variance associated with plaques and an error associated within runs. The plaque error includes experimental error and variations within individual plaques. The error associated with runs includes these errors and variations from plaque to plaque. There are five (5) replicated runs (1-5) and four (4) not replicated which provide an estimate of error from run to run.

Table Number I tabulates the analysis of variance data necessary.

From the standpoint of production, the estimate of sigma in Table Number I that is most meaningful is the "within runs and replicates" in each. This indicates the variability to be expected when fresh start-ups will take place. The coefficient of variation (est. of sigma/mean as %) is a useful index. These are:

<u>ITEM</u>	<u>WITHIN RUNS &amp; REP. COEF. OF VARIATION</u>	<u>WITHIN PLAQUES COEF. OF VARIATION</u>
Strength	= 15.4%	= 13.3%
Void	= 5.10%	= 4.4%
Thickness	= 4.11%	= 3.5%
Weight	= 2.46%	= 2.1%

To reduce these variations, other controls would have to be instituted than were used during these runs. These data can serve as a baseline for any change made. It is interesting to note that approximately 85% of the variation is within an individual plaque.

TABLE NUMBER I  
 ANALYSIS OF VARIANCE (ERROR ANALYSIS)

(1) Response: Strength (Mean = 507.9)

	<u>DF</u>	<u>SSQ</u>	<u>MS</u>	<u>S</u>
Total	749	22745646.	30368.	174.3
Within Plaques	666	3025477.	4543.	67.4
Within Runs	736	4276397.	5810.	76.2
Within Runs & Reps.	741	4508849.	6085.	78.0
Between Runs & Reps.	8	18236798.	2279600.	1510.

(2) Response: Void (Mean = .02295)

Total	749	.002383	.000003181	.00178
Within Plaques	666	.0006688	.00001004	.001002
Within Runs	736	.001004	.000001364	.001167
Within Runs & Reps.	741	.001014	.000001368	.001170
Between Runs & Reps.	8	.001369	.0001711	.01308

(3) Response: Thickness (Mean = .02876)

Total	749	.002369	.000003162	.1778
Within Plaques	666	.0006796	.000001020	.001010
Within Runs	736	.001022	.000001388	.001178
Within Runs & Reps.	741	.001034	.000001396	.001182
Between Runs & Reps.	8	.001334	.0001668	.01291

(4) Response: Weight/2 sq.in. (Mean = 1.6935)

Total	749	1.4381	.001919	.04381
Within Plaques	666	.8386	.001259	.03548
Within Runs	736	1.1512	.001564	.0395
Within Runs & Reps.	741	1.2812	.001729	.04158
Between Runs & Reps.	8	.1569	.01961	.1400

LEGEND

DF = Degrees Freedom

SSQ = Sum of Squares

MS = Mean Square = SSQ ÷ DF

S = Sigma = Square Root of MS

#### D. Multiple Regression Analysis-Development of Equation

To evaluate the effect of the independent variables in the matrix, multiple regression analyses were made with and without all possible first order interactions among the independent variables. Responses were strength, void and thickness. The independent variables are:

1. Hot Zone Temperature
2. Belt Speed
3. Dewpoint of Atmosphere
4. Atmosphere Amount
5. Plaque Spacing
6. First Cooling Chamber Water Temperature
7. Second Cooling Chamber Water Temperature
8. Plaque Sequence
9. Weight of 2 sq.in.

The regression equations were derived with and without interactions and the models (equations) are presented in the Appendix. After derivation of the regression equations, they were checked against the known data points, that is, experimental runs 1 through 8 of the fractional factorial and Run 9, the mid-level run.

The data were analyzed by a multiple regression computer program (Gardner Program discussed previously) both with and without inclusion of interactions among the independent variables. The independent data matrix was first formed (interactions formed) as raw data except temperature in which the decimal was moved 3 digits. Following this, each vector was coded with maximum equal to +1, and the minimum equal to -1, and all values were prorated between these extremes. The variables are numbered identically to the list above. The interactions

are shown in Exhibit No. 1 in the Appendix. These variables "numbers" correspond to the "x" numbers in the Appendix, Exhibits 2 - 4 of the models (equations) for each response (strength, void and thickness).

The regressions were solved without and with interactions and the "value" of the more complicated model can be judged by comparing the squared multiple correlation coefficients for each. These are listed below in Table No. II.

TABLE NUMBER II

MULTIPLE CORRELATION COEFFICIENTS  
(SQUARED) FOR VARIOUS MODELS

	<u>R<sup>2</sup></u>
Strength, No Interactions	.746
Strength, With Interactions	.850
Void, No Interactions	.554
Void, With Interactions	.607
Thickness, No Interactions	.551
Thickness, With Interactions	.605

This coefficient is 1 minus the ratio of the residual sum of squares over the total sum of squares. The "sum of squares" is a measure of the variability of the data so this coefficient measures the "amount" accounted for by the model (variables). There is an appreciable increase for the interaction models over that including no interactions.

The other statistical terms have been referenced earlier. When the coefficients are small, the program prints them in "E" format at the right side. The sign and integers following the "E" is the logarithmic characteristic and shows the digits to shift the decimal (- to the left; smaller). To graphically show a model effectiveness,

Figure 1 shows a plot of the residuals (prediction from model - actual observation) versus the observation. A random scatter shows no variables predominately causing effects. This also shows the values of the residuals resulting from the 750 observations originally used to setup the models.

Having quite complex models such as these, it is relatively easy to create a vector among the interactions which is beyond that found in the original data maxima or minima. Obviously in creating the matrix for predicting some set of values, the same maxima and minima must be used as was used to create the model. If some vector exceeds these limits either positively or negatively, an extrapolation may result. Since the model is very complex, this extrapolation may result in a predicted response that has a large error associated with it. Besides giving a response quite different from that expected, the standard error of the prediction ( $\bar{Y}$ ), which the program prints for each, will be substantially greater than the original data values. By printing the values for the  $X$  matrix after coding, the extrapolated vector can readily be located.

PLOT OF RESIDUALS VS. OBSERVATIONS

FOR STRENGTH (9 VAR. MATRIX)

W/INT)

NOTE: A = 1 VALUE, B = 2 VALUES,  
C = 3 VALUES \* > 26 VALUES, THE SAME VALUE

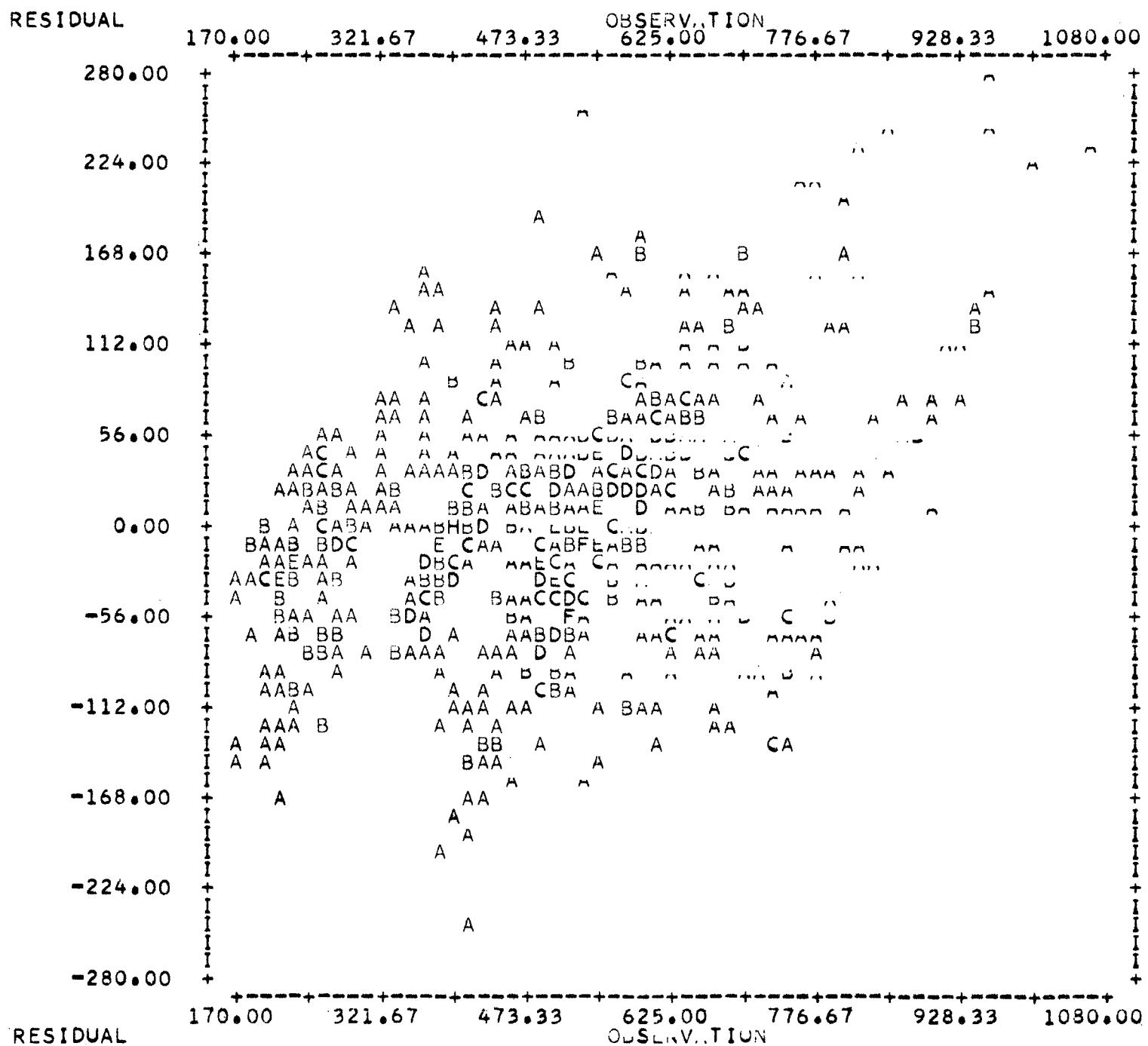


FIGURE NUMBER 1

E. Use of the Regression Equation (See Appendix, Exhibits 2-4)

The derived equation (mathematical model) is very useful.

This equation can be used to:

- (1) Assess Variability Within the Process
  - (2) Predict Plaque Characteristics for Various Furnace Settings
  - (3) Optimization of the Process
- (1) Assess Variability Within the Process

This is done by programming maximum and minimum tolerance levels for each of the independent variables into the computer. Each variable is held constant except the one being investigated which can be varied high and low to determine the effect of changing this particular variable. In turn, all of the independent variables can be analyzed to determine the effect upon the process. This analysis has been performed for the 9 independent variables at various settings of the furnace and various changes of the other independent variables. The nominal settings and the amount of variation programmed into the computer are shown in the following table.

TABLE III  
ANALYZED VARIATION FOR THE  
PLAQUE MANUFACTURING INDEPENDENT VARIABLES

<u>INDEPENDENT VARIABLE</u>	<u>NOMINAL SETTING</u>	<u>RANGE OF VARIATION</u>
Hot Zone Temperature	1760°F, 1860°F	30°F
Belt Speed	6 in/min	1 in/min
Dewpoint	25°F, 35°F	10°F
Atmosphere Amount	800 cfh	20 cfh
Spacing	0.1"	1.8"
Cooling Zone 1	180°F, 75°F	16°F
Cooling Zone 2	83°F, 100°F	16°F
Plaque Sequence	6	2
Weight of 2 sq.in.	1.757 gm, 1.920 gm	0.24 gram

The computer run was made and the following conclusions were made for this variability study:

1. An increase of 30°F in temperature at 1760°F causes a 2% decrease in void and thickness and an increase in strength of 12%. However, this same increase at 1860°F can cause no change in void or thickness and can cause diminished strengths of 5-6%.
2. An increase in belt speed of 1 inch/minute will cause less than a 1% change in void and thickness. Strength changes of less than 2% were shown at 1760°F but a 6% strength decrease was predicted at 1860°F with this increase in belt speed.
3. Dewpoint exhibits especially strong affects for only a plus or minus change of 5°F. Thickness and void changes of 5% were seen at 1860°F and strength changes of up to 40%. Dewpoint effects are most pronounced on the lightest weight plaques.
4. A change of  $\pm$  10 cfh of atmosphere amount causes no change in void or thickness and less than 1% in strength.
5. Close spacing is most desirable for uniform plaques.
6. A 16°F increase in temperature of the first cooling zone water increases thickness and void in the range of 0.0 to 0.7%, and increases strength approximately 2.5%.
7. A 16°F increase in temperature of the second cooling zone water predicted large increases in strength and significant changes in void and thickness.
8. A plaque weight increase obviously increases thickness and void. Strength is increased at 1760°F as plaque weight is increased, but at 1860°F strength is decreased as plaque weight is increased.

NOTE: Supporting data are presented in the Appendix as Exhibit 5.

(2) Predict Plaque Characteristics for Production Settings

The equation can be used to predict plaque characteristics for a production run. In this manner, certain equipment limitations could be overcome by investigation and computation of the alternative furnace settings. This would lead to optimization of the process.

(3) Optimization of the Process

The complex interactions have shown that the process should be optimized by analysis of existing data. For example, it was shown that a plaque weight increase resulted in a stronger plaque at 1760°F, but a weaker plaque at 1860°F.

It is planned to perform this optimization process similarly in technique to the variability assessment described in (1) above. Variations of each independent variable will be programmed larger than those used in the above analysis. Also, more test data are being accumulated. These data will be added to the data of Runs 1-9 to improve the regression equation and to reduce error in the analysis (shown as S. E. (Y)). These data are being made available from production runs and other independent furnace investigations. Additional data will be accumulated on changes of dew-point and changes of cooling zones.

### III. IMPREGNATION-POLARIZATION EXPERIMENT

#### A. IMPREGNATION-POLARIZATION PROCESS

The impregnation-polarization process is the means by which the active material is applied in a porous nickel plaque. This step is accomplished by soaking under vacuum a porous nickel plaque in either nickel or cadmium nitrate solutions. In order to convert the nitrates to the desired hydroxides, the plaques are immersed in either potassium hydroxide (KOH) or sodium hydroxide (NaOH) and a cathodic polarizing current applied.

The impregnated plaques are then washed and dried. This process is repeated until the plaques are impregnated with the desired amount of active material.

#### B. EXPERIMENTAL DESIGN

The designed experiments for the impregnation of polarization study will be conducted to obtain information on the effect of production parameters and error variance. This design, like that used for the raw plaque study previously reported, is a partially replicated fractional factorial of a sequential series type using twenty-one (21) variables and three (3) responses. The variables, their designation and levels are listed in Tables IV and V.

TABLE IV  
POSITIVE IMPREGNATION STUDY

<u>Variables</u>	<u>Designation</u>	<u>Levels</u>
1	Specific Gravity of Nickel Nitrates	1.700 - 1.800
2	Free Acid	1 - 4 gm/liter
3	Temperature of Nitrate	140°F - 200°F
4	Time of Impregnation	15 minutes - 1 hour
5	Vacuum	0 - 15 Inches
6	Wash Time	10 Minutes - 30 Minutes
7	Wash (Number of Cycles)	1 - 3
8	Wash Water Temperature	R.T. - 150°F
9	pH of Wash Water	Measured
10	Type of Caustic	KOH - NaOH
11	Specific Gravity of Caustic	1.200 - 1.300
12	Temperature of Caustic	80°F - 150°F

TABLE IV (Cont.)  
POSITIVE IMPREGNATION STUDY

<u>Variables</u>	<u>Designation</u>	<u>Levels</u>
13	Amount of NH <sub>3</sub> in Caustic	Measured
14	Amount of CO <sub>2</sub> in Caustic	Measured
15	Amount of OH in Caustic	Measured
16	Polarization Current	.1 - .4 amps/sq.in.
17	Polarization Time	15 Minutes - 1 Hour
18	Voltage of Plaque to Ref. Electrode	Measured
19	Amount of Cycles with Same Caustic	1 - 5
20	Number of Total Cycles	Measured
21	Type of Plaque	3 Types

Responses

1. Pick-up Weight
2. Capacity
3. Plate Testing (Characterization) These tests will be detailed in the Third Quarterly Report.

TABLE V  
NEGATIVE IMPREGNATION STUDY

<u>Variables</u>	<u>Designation</u>	<u>Levels</u>
1	Specific Gravity of Cadmium Nitrate	1.800 - 1.900
2	Free Acid	.2 - .5 gm/liter
3	Temperature of Nitrate	110°F - 140°F
4	Time of Impregnation	15 Minutes - 1 Hour
5	Vacuum	0 - 15 Inches
6	Wash Time	10 Minutes - 30 Minutes
7	Wash (Number of Cycles)	1 - 3
8	Wash Water Temperature	R.T. - 150°F
9	pH of Wash Water	Measured
10	Type of Caustic	KOH - NaOH
11	Specific Gravity of Caustic	1.200 - 1.300
12	Temperature of Caustic	80°F - 150°F
13	Amount of NH <sub>3</sub> in Caustic	Measured
14	Amount of CO <sub>2</sub> in Caustic	Measured
15	Amount of OH in Caustic	Measured
16	Polarization Current	.1 - .4 Amps/sq.in.
17	Polarization Time	15 Minutes - 1 Hour
18	Voltage of Plaque to Ref. Electrode	Measured
19	Amount of Cycles with Same Caustic	1 - 5
20	Number of Total Cycles	Measured
21	Type of Plaque	3 Types

Responses

1. Pick-up Weight
2. Capacity
3. Plate Testing (Characterization)  
These tests will be detailed in the Third Quarterly Report.

All the variables will be measured and the actual numbers, not the design levels, will be used in the analysis.

Three types of raw plaque material have been selected for impregnation. One is a very low strength, high porosity plaque produced at a low sintering temperature. The second and third groups are successively stronger plaques. These plaques will be used for both positive and negative experiments. The settings for these plaques were selected as a result of the plaque investigations conducted during the first quarter. Complete data for these plaques will be included in the Third Quarterly Report.

All three types of plaques will be impregnated to the above mentioned program plan. The data gained from these plaques will be used to complete the analysis for the raw plaque study conducted during the first quarter of this program.

#### C. Experimenting Equipment

The equipment used for the impregnation/polarization study is an exact scale down of the production equipment. Figures 2 and 3 show the vacuum impregnation tanks, nitrate/caustic storage tanks and power supplies. The vacuum tank is equipped with a vacuum gage, thermocouples, voltage taps, power connectors and suction fan. The storage tanks have independent, controlled induction heating systems under the thickly insulated side walls. Current for polarization is supplied by three (3) Rapid Electric Power Supplies (Model S-550). All water used during the experiments will be supplied by a Model HB-455-1 Deionizing Unit produced by Illinois Water Treatment Company (See Figure 4). This unit has a 350 gal/hr. rate with

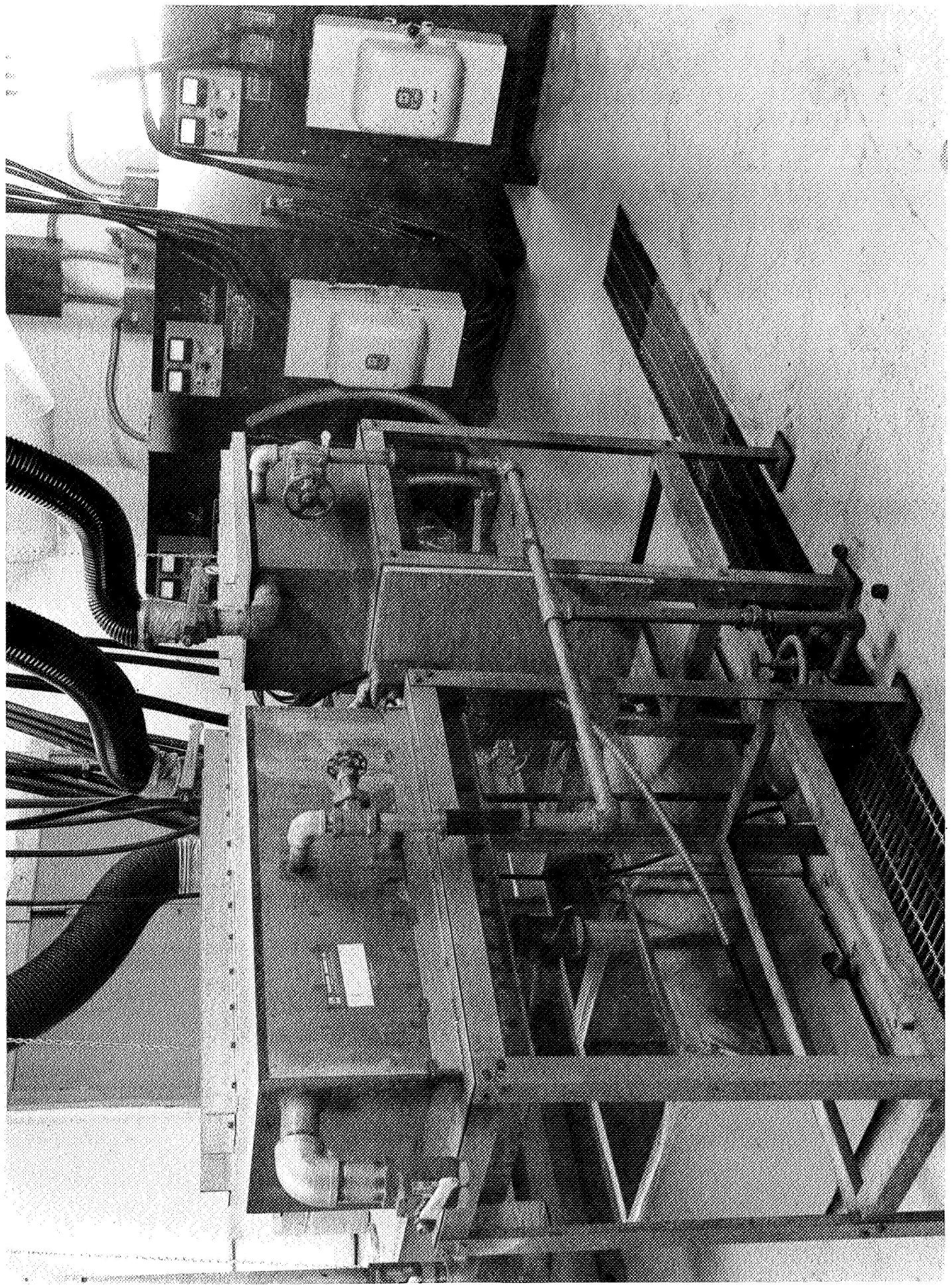


FIGURE 2

- 15A -

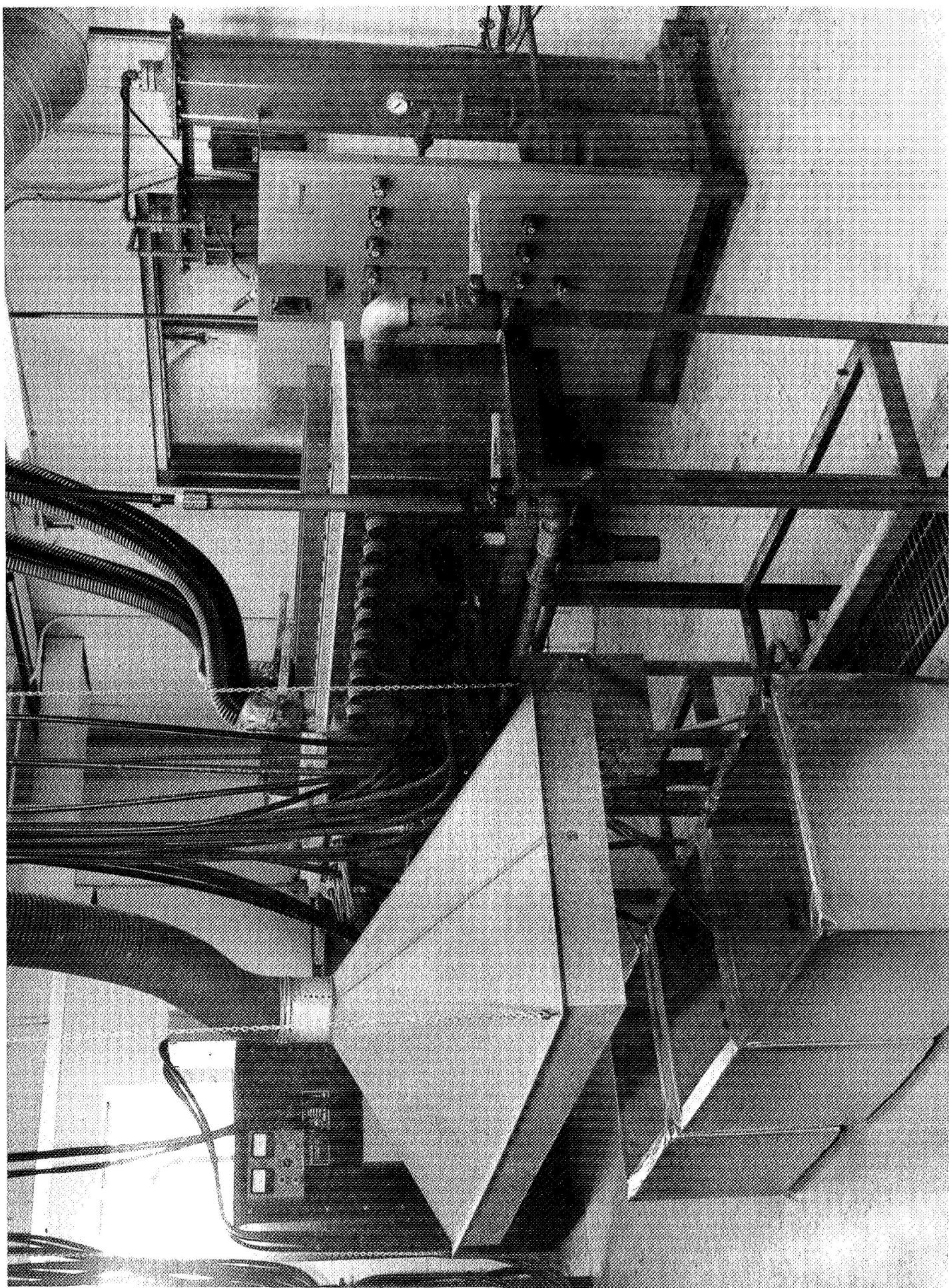


FIGURE 3

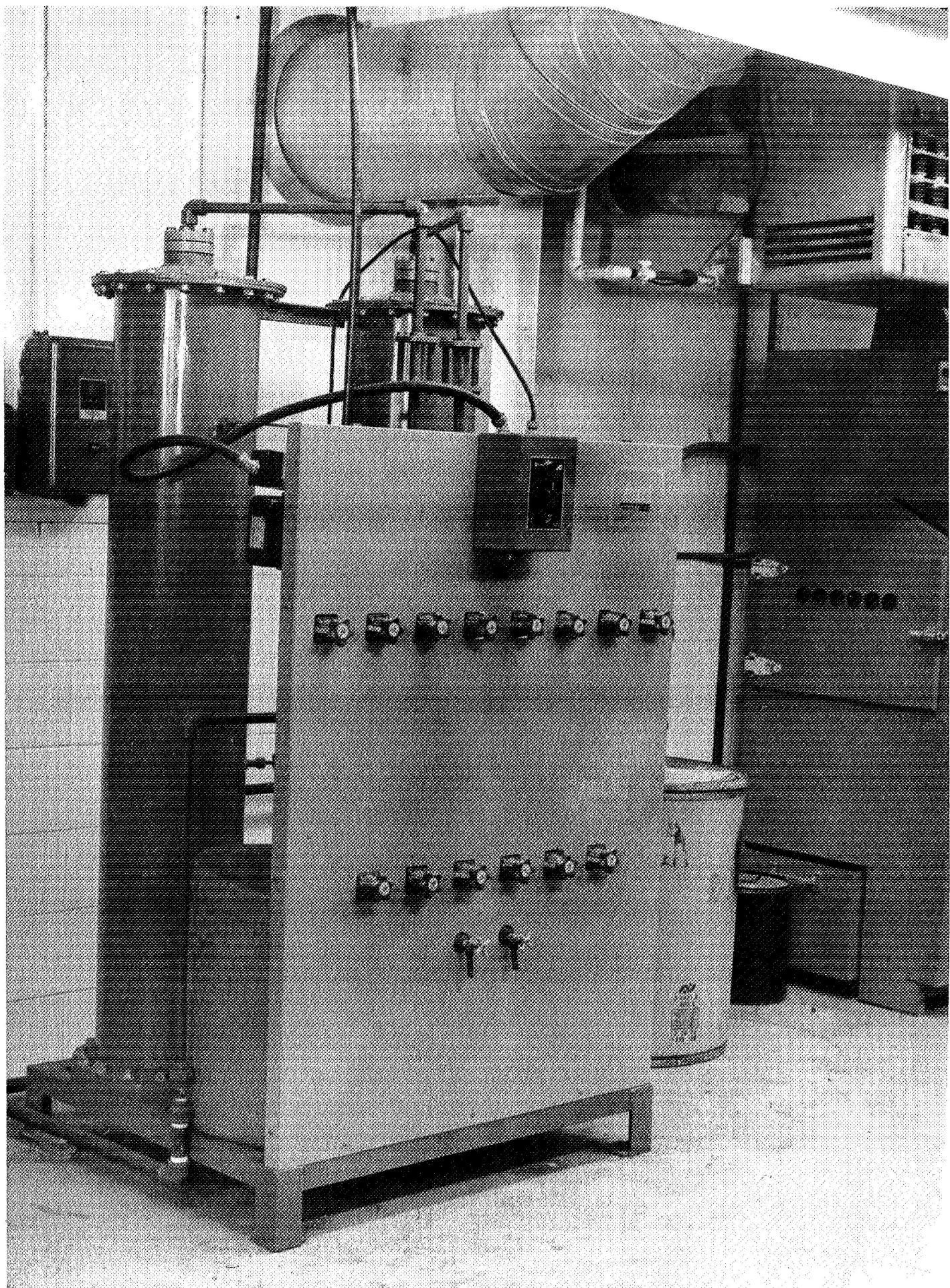


FIGURE 4

- 15C -

a 40,000 gallon capacity at a purity level of less than 3 ppm.

Laboratory equipment has been assembled to conduct the required chemical analysis on the nitrates and caustic used.

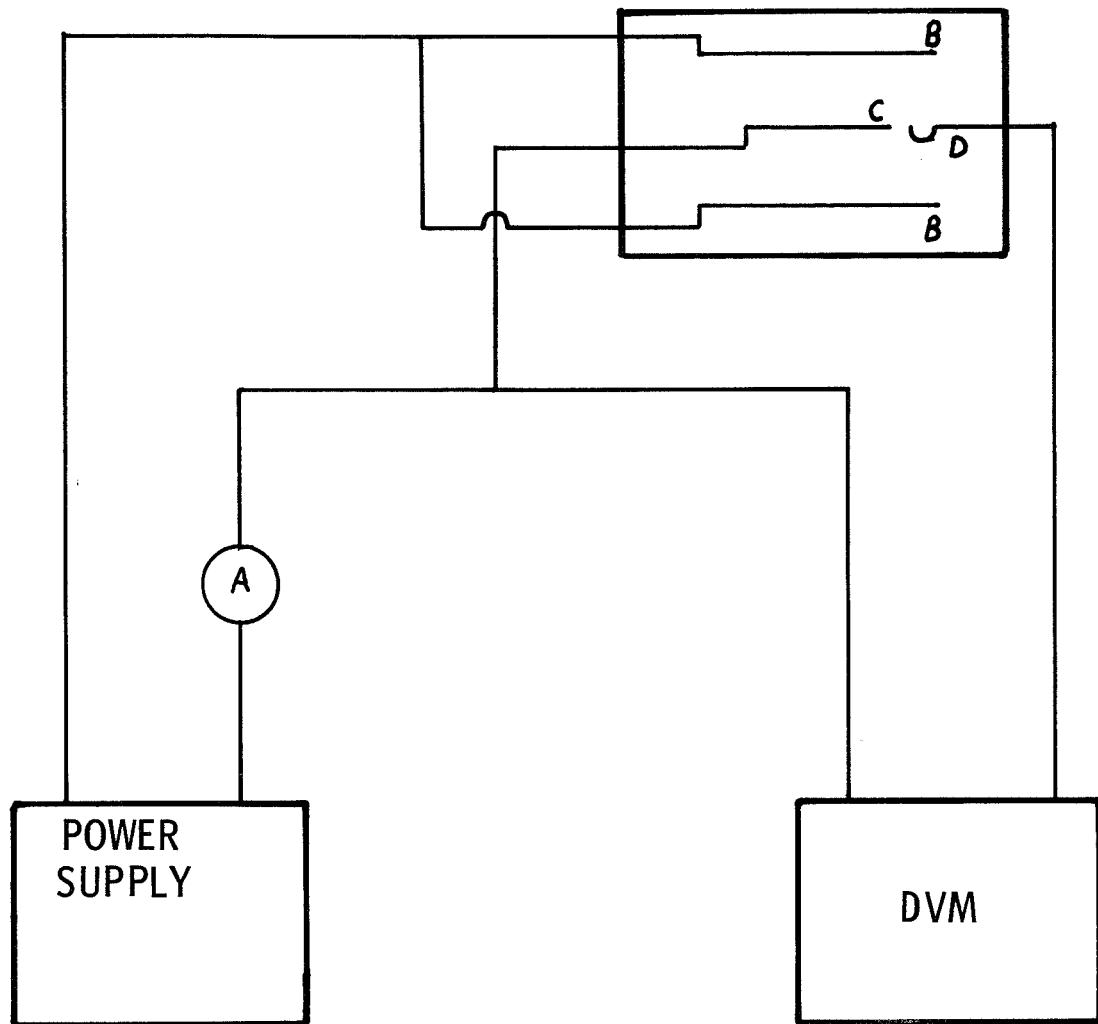
D. Experiment Testing and Control

In order to maintain the variables at their proper levels, as shown in the program plan, it will be necessary to measure certain properties. Although an effort will be made to hold the desired levels, the actual measurement will be used in the regression analysis.

Specific gravity of the solutions will be measured with hydrometers. Free acid of the nitrates will be measured using the titration method of Dr. A. Fleischer<sup>1</sup>.

Temperatures will be read with thermometers and thermocouples. Vacuum can be measured with the vacuum gauge installed in the tank cover. Polarization currents will be measured using shunts on the line for each power connector going to the tank as can be seen in Figure 3. Chemical analysis of the nitrates and caustic will be conducted to measure the amounts of OH, CO<sub>2</sub> and NH<sub>3</sub>. Chemical analysis of the OH content will be by titration. The standard double end point titration method will be used for CO<sub>2</sub> content. To determine the NH<sub>3</sub> content, a modified Kjeldahl distillation-titration method using 4% boric acid will be used.

Each plaque and subsequent plates will be weighed to determine the amount of active material picked up in the porous nickel plaque. A dimensional analysis will be done on each plaque before and after impregnation to determine its dimensional stability. Electrical capacity tests will be conducted on the test samples to measure the actual, useable active material. These electrical tests will be run on test samples using a Hg/HgO reference electrode and will be connected as shown in Figure 5. An attempt will be made to use the Four Point Bend Test



- A Ammeter
- B Sintered Nickel Counterelectrodes
- C Test Sample
- D Hg/HgO Reference Electrode

SINGLE PLATE TEST SCHEMATIC

FIGURE 5

before and after impregnation to determine the affect of impregnation  
on the physical strength.

#### IV. BATTERY SEPARATOR STUDY

The first task in this study is to develop accurate and predictable methods for screening non-woven nylon separator materials. Test methods used by various people and organizations connected with the battery industry have been investigated. The test methods used or recommended by the NASA Interim Specification<sup>2</sup>, Fleischer & Cooper<sup>3</sup>, Dr. T. King<sup>4</sup>, Pellon Corporation<sup>5</sup>, the GAC-OAO Specification<sup>6</sup> and Eagle-Picher are listed in Table VI.

The study thus far has been confined to the non-woven nylon type separator materials. The final objective of this study is to obtain testing methods which will give the most accurate and repeatable results to be used in screening separator material.

Future work will include the comparison of different testing methods on the same lot of separators and the feasibility of testing separator materials under compression.

A procurement and testing specification will be prepared for non-woven nylon separator material during the third quarter.

TABLE NUMBER VI - COMPARISON OF SEPARATOR TEST PROCEDURES

TEST	EAGLE-PICHER	NASA INTERIM SPRC <sup>2</sup>	FLEISCHER & COOPER <sup>1</sup>	KING, LACKNER & BAINES <sup>4</sup>	PELTON CORP. <sup>3</sup>	OAO SPEC. <sup>6</sup>
1. IDENT. OR TRACEABILITY	Lot & Roll Number	Basic Fiber & Separator Mfg., Part, Lot & Style No., Condition (Washed or Treated)			Fiber Mfg., Part & Lot No., Date of Mfg.; Separator Mfg. Style & Lot No.; Date	
2. WEIGHT	Weigh One Square Yard	1 m <sup>2</sup> Approx. Shape			6 ea. 6.5 cm x 2.5 cm Samples	
3. THICKNESS	Cady Gage DW1	Ames Model 262 Platform Dial Micrometer	ASTM Test	Cady Gage DW1	5 samples 12.7 x 2.5 cm; 3 in/min. grip separation of 3 inches repeat after 24 hrs. @ 70°C Ames Model 262; Also variation in thickness is recorded.	
4. TENSILE STRENGTH	1" Wide Sample 3/min. pull with fiber direction (Dev. Tests)	Both directions, Also % elongation @ break	2" x 3" Sample @ 4 in/min; Run both wet & dry	2" wide Sample	1" x 6" Sample, Immense 1" x 6" Sample	5 samples 12.7 x 2.5 cm; 3 in/min. grip separation of 3 inches repeat after 24 hrs. @ 70°C
5. WICKING RATE	1" x 6" Sample, Immense 1/4" in KOH; Record Time to 1/2" & 1" (Dev. Tests)	Recorded.	ASTM Test, Before & After a 1 hr. boil in 30% KOH	Immense 1/4" in KOH; Record Time to 1/2" & 1"	1/4" in KOH, Record Time to 1"	
6. AIR PERMEABILITY			ASTM Test, D3774-61	ASTM Test, D3774-61	ASTM Test, D3774-61	
7. ELECTROTE ABSORPTION	4" x 4"; 1 hr. Soak 10 Second Drip, Record Weight	4 Hr. Covered Soak; wipe across lucite block, record weight.	3/4" x 1 1/4" sample; 5 min. soak, wipe on lucite block; record weight.	1" x 6" Sample; Soak 10 min., 3 min. drip, record weight	1" x 6" Sample; Soak 10 min., place between .010 plateau same size as sample pressure .1 oz/in <sup>2</sup>	6 each samples, 6.5 x 2.5 cm; weigh & measure, soak 3 hrs, record weight & dimensional changes, soak 1 more hour.
8. ELECTROLYTE RETENTION		Sample from absorption drained 15 min. @ 45° angle on lucite.	Centrifuge @ 25 g for 2 min.			After second soak above, wipe on lucite plate and reweigh.
9. ELECTRICAL RESISTANCE	4" x 4" sample between electrodes, 10 amps applied; measure voltage drop DC	D.C. method essentially same as Fleischer	Cd Electrodes, D. C., Ref. Ig/HgO Electrode, using a null potentiometer			Modified Fleischer Method; Cd/Cd(OH) <sub>2</sub> Electrode; D.C., Hg/HgO Ref. Electrode
10. WETTABILITY	Time to wet out, visual.	Measurement of "R" @ 5 second intervals.				
11. RESISTIVITY	From "R" & Thickness					From "R" & Thickness
12. H <sub>2</sub> O SOLUBLES	Foaming Test on Shaking in H <sub>2</sub> O					
13. ORGANIC LEACHABLES	10 cm <sup>2</sup> sample stirred overnight in methylene chloride; evap.; IR on residue.					
14. POROSITY	KMnO <sub>4</sub> Tests (Dev.)	10 cm <sup>2</sup> sample stirred overnight in methylene chloride; evap.; IR on residue.	5 Hr. Trichloroethylene Soxlett Extraction	5 Hr. Trichloroethylene Soxlett Extraction	4" x 4" sample; dry @ 104°C extract with H <sub>2</sub> O; 4-5 hrs. Dry, Weigh, Record Wt. Loss	3 samples, 10 cm <sup>2</sup> , methane extraction; weigh residue & run IR trace.
15. OXIDATION RESISTANCE		10 cm <sup>2</sup> sample stirred overnight in methylene chloride; evap.; IR on residue.	1. H <sub>2</sub> O Solubles 200 ml 0.1N KOH, Titrate w/0.001N KMnO <sub>4</sub>	1. H <sub>2</sub> O Solubles 200 ml 0.1N KOH, Titrate w/0.001N KMnO <sub>4</sub>	1. H <sub>2</sub> O Solubles 200 ml 0.1N KOH, Titrate w/0.001N KMnO <sub>4</sub>	Same as NASA, Inorganic Analysis subject to approval
16. INORGANIC CONTENT	Test for Carbonate, Zinc Chloride, Nitrate, Nickel		2. Sample boiled 100 ml 5% KMnO <sub>4</sub> 1 hr. (Wt. Loss Test)	2. Sample boiled 100 ml 5% KMnO <sub>4</sub> 1 hr. (Wt. Loss Test)	Burst Strength ASTM D1117-63 by Request	C of C for antistatic agent or wetting agents.
17. OTHERS						

## V. CERAMIC-TO-METAL SEAL COVER ASSEMBLIES

There are two pertinent reports that have been prepared regarding ceramic-to-metal terminal cover assemblies for hermetically-sealed cells which are especially useful for nickel-cadmium cell designers and battery users. These reports are "A Study of the State-of-the-Art of Hermetic Seals for Secondary Alkaline Spacecraft Cells" (Final Report NAS5-10432) (Ref. 7). by Dr. Willard Scott of TRW and the paper presented by Mr. R. A. Steinhauer (Ref. 8) of Hughes Aircraft Company at the NASA/GSFC Workshop on November 17, 1970. It is recommended that all battery cell designers and cell and battery users be familiar with these reports as they contain excellent descriptions of current state-of-the-art and important design parameters that must be considered for the development of high reliability ceramic-to-metal seals. The following discussion presents specific areas and recommendations based on Eagle-Picher experience and are submitted to complement the above two reports.

A specification for quality control of ceramic-to-metal terminal cover assemblies for hermetically-sealed nickel-cadmium cells is being prepared. Source Control Drawings will be prepared for each manufacturer which also contain provisions for source control of component parts. This Source Control Drawing will be complemented with a Quality Assurance Specification with the format as outlined below. Ceramic assemblies will be procured using these specifications and drawings for deliverable cells on this contract.

The metal parts used in the ceramic-to-metal seal consist of the cell cover, terminals, stress relief collars, and braze cups or caps.

The cell cover is manufactured of stainless steel with the preferred material being 304L per QQS-766 Condition A. Type 304 stainless steel has also been satisfactorily used. The only major difference in the two materials is the percentage of carbon which is 1/2% greater in Type 304.

Terminals are manufactured from stainless steel or nickel. Nickel is generally preferred by Eagle-Picher due to its better conductivity. Also, the machineability of stainless steel presents some problems. Nickel 270, as supplied by International Nickel, is preferred. This nickel is the purest nickel commercially available. Nickel 200 has been used, but porosity problems of this nickel have led to leak problems of ceramic to metal seals. International Nickel was consulted on this problem and recommended the use of Nickel 270 to eliminate the fissures and porous areas caused by impurities in the less pure grades of nickel.

Stress relief collars and upper braze cups, also referred to as sleeves and caps, can be manufactured from pure nickel or nickel-iron alloy 42 or 52. The use of nickel is not recommended for the active metal system where there is contact with titanium, as local nickel alloying with titanium occurs. Alloy 42 is preferred for the active metal system because of its closer matching thermal expansion coefficient to that of the ceramic insulator. However, alloy 42 parts must be copper plated to improve wettability during the brazing operation. Alloy 52 does not need to be copper plated and appears to be a good compromise for these parts; however, the nickel iron alloys are not corrosion resistant and must be protected from humidity exposure to prevent corrosion and eventual leakage problems.

Ceramic insulators are available in many varieties of alumina. The 94 to 95% alumina content ceramic bodies have been used in most nickel cadmium spacecraft batteries to date. Ceramic bodies are available from several vendors in the general class of 94% minimum, 99% minimum, and ultra-high purity varieties of alumina such as Coors AD999 and Vistal, or G. E. Lucalox. It is difficult to assess the improvement to be gained from higher alumina content ceramic although leakage paths (cracks) at the ceramic braze interface have been shown.<sup>7</sup> The braze interface area contains fused glass phases. It follows then that removal of the silica content from the ceramic insulator would be desirable. This would be true except that the metallizing material and its formula must be changed with different alumina contents. With the molymanganese system, silica must be added to the metallizing agent; with the active metal system, we are advised that no glassy phases need be added in the metallizers. The composition of the metallizing layer or the metallizing material is generally proprietary.

The braze alloy employed has been the silver copper eutectic for many space type units. Also Ceramaseal, Inc. has incorporated approximately 5% palladium in their brazing alloy for a number of space qualified seals to improve corrosion resistance. This results in a composition of 27% copper, 68% silver, and 5% palladium.

Preliminary work has begun on a quality assurance specification for the manufacture of ceramic-to-metal seals. This specification is divided into the following areas:

I. Materials Testing

- A. Metal Parts
- B. Ceramic Insulators
- C. Brazing and Metallizing Materials

II. Cleaning and Handling

- A. Cleaning Metal Parts
- B. Cleaning Ceramic
- C. Handling
- D. Storage

III. Manufacturing

- A. Brazing Operations
- B. Welding

IV. Testing

- A. Start Up - Qualification
- B. 100% Testing

V. Packaging

## VI. CONCLUSIONS AND RECOMMENDATIONS

1. Analysis of variance has shown that 85% of the total variation occurs within any individual plaque. It is further concluded that a 2% variation is introduced into the plaque by virtue of the manufacturing process.
2. It is recommended to pursue additional work to reduce the 2% variation inherent in the process. This work should concentrate on uniformity improvement of individual plaques.
3. It is concluded that there are strong interactions in the plaque manufacturing process.
4. Temperature, belt speed and dewpoint effects are non-linear due to interactions.
5. Hot zone temperature, belt speed, dewpoint, weight and second cooling zone temperature cause relatively strong effects on plaque characteristics.
6. Relatively high degrees of control (with respect to amount of change) are available for atmosphere amount, belt speed and spacing.
7. Improved controls are recommended for temperature of hot zones and cooling chambers, and control of dewpoint.
8. It is also recommended and planned to conduct further optimization studies for manufacturing nickel plaques.

## VII. PROGRAM PLAN

The following areas will be investigated during the third quarter of the contract.

1. Impregnation, polarization, and formation studies will be continued.
2. Based on the test results from the first experimental runs, it is planned to alter the factorial design experiment to provide data with less total time required to perform the experiments. It will be necessary to reduce the number of experiments in order to do the work in a timely manner.
3. Additional tests will be performed on the three (3) runs of unimpregnated plaque which are being used for the impregnation study.
4. The separator specification shall be prepared.
5. The plaque quality assurance specification will be finalized.
6. Ceramic-to-metal seal source control drawing will be prepared for the cover assemblies to be purchased for the contract.
7. The cell test study will be formally planned.

REFERENCES

- <sup>1</sup>Dr. A. Fleischer, Free Acid Titration in Impregnating Solutions, Nickel or Cadmium Nitrates, Report No. 68AF11E01, November 25, 1968.
- <sup>2</sup>NASA Interim Model Specification S-716-P-23 for High Reliability Nickel-Cadmium Spacecraft Cells dated 30 April 1969.
- <sup>3</sup>J. E. Cooper & A. Fleischer, Characteristics of Separators for Alkaline Silver Oxide Zinc Secondary Batteries, Screening Methods, Report No. PB166780.
- <sup>4</sup>Dr. T. E. King, Q. C. Procedures and Measurements - Polypropylene Separators For Sealed Nickel-Cadmium Batteries, Proceedings at the Symposium on Battery Separators, The Columbus Section of The Electrochemical Society, February 18-19, 1970. (T. E. King, J. L. Lackner & R. L. Haines).
- <sup>5</sup>Pellon Alkaline Battery Separator Material Description of Test Methods dated October, 1968.
- <sup>6</sup>GAC Specification AV-252CS-25E for Nickel-Cadmium Storage Cells Power Supply Subsystem, Orbiting Astronomical Observatory (OAO), dated 4/15/69.
- <sup>7</sup>Dr. W. R. Scott, Final Report, A Study of The State-of-the-Art of Hermetic Seals for Secondary Alkaline Spacecraft Cells, (June 20, 1967 - 20 March 1968) NAS5-10432.
- <sup>8</sup>R. A. Steinhauer, Design of an Improved Long Life Ceramic-to-Metal Seal for Spacecraft Nickel-Cadmium Cells, presented at the NASA/GSFC Battery Workshop November 17, 1970.

A P P E N D I X

EXHIBIT 1

VARIABLE NUMBERS FOR CORRESPONDING INTERACTIONS  
USING A 9 VARIABLE MATRIX

```
D(47)=D(11)
D(01)=D(01)*.001
D(10)=D(01)*D(01)
D(11)=D(01)*D(03)
D(12)=D(01)*D(04)
D(13)=D(01)*D(05)
D(14)=D(01)*D(06)
D(15)=D(01)*D(07)
D(16)=D(01)*D(08)
D(17)=D(01)*D(09)
D(18)=D(02)*D(03)
D(19)=D(02)*D(04)
D(20)=D(02)*D(05)
D(21)=D(02)*D(06)
D(22)=D(02)*D(07)
D(23)=D(02)*D(08)
D(24)=D(02)*D(09)
D(25)=D(03)*D(04)
D(26)=D(03)*D(05)
D(27)=D(03)*D(06)
D(28)=D(03)*D(07)
D(29)=D(03)*D(08)
D(30)=D(03)*D(09)
D(31)=D(04)*D(05)
D(32)=D(04)*D(06)
D(33)=D(04)*D(07)
D(34)=D(04)*D(08)
D(35)=D(04)*D(09)
D(36)=D(05)*D(06)
D(37)=D(05)*D(07)
D(38)=D(05)*D(08)
D(39)=D(05)*D(09)
D(40)=D(06)*D(07)
D(41)=D(06)*D(08)
D(42)=D(06)*D(09)
D(43)=D(07)*D(08)
D(44)=D(07)*D(09)
D(45)=D(08)*D(09)
D(46)=D(01)*D(02)
```

\*SYMBOL \* = MULTIPLY  
THUS = D (15) = D (01) \* D (07)  
MEANS VAR.(15) = VAR. (01) x VAR. (07)

EXHIBIT 2

MODEL FOR THICKNESS  
USING INTERACTIONS

X	RSQR	X	B	COEF	SE(B)	T	ANALYSIS OF Y	1
1	0.9995		0.0039	0.0024	1.62	0.397815E-02		
2	0.9985		0.0020	0.0011	1.77	0.200483E-02		
3	0.9997		0.0015	0.0045	0.33	0.151467E-02		
4	0.9999		0.0063	0.0051	1.22	0.635395E-02		
5	0.9994		0.0024	0.0017	1.35	0.243882E-02		
6	0.9999		-0.0134	0.0056	-2.38	-0.134953E-01		
7	1.0008		0.0067	0.0030	2.25	0.676428E-02		
8	0.9993		-0.0040	0.0024	-1.69	-0.408308E-02		
9	0.9899		0.0015	0.0015	0.95	0.150703E-02		
11	0.9998		-0.0033	0.0070	-0.48	-0.338785E-02		
12	0.9999		-0.0060	0.0054	-1.11	-0.604179E-02		
14	0.9999		0.0123	0.0054	2.28	0.123480E-01		
15	0.9997		-0.0143	0.0051	-2.77	-0.143094E-01		
16	0.9979		0.0041	0.0014	2.86	0.418990E-02		
17	0.9988		-0.0013	0.0028	-0.45	-0.130762E-02		
18	0.9984		0.0018	0.0016	1.16	0.189183E-02		
19	0.9953		-0.0022	0.0008	-2.53	-0.224552E-02		
20	1.0121		0.0025	0.0004	5.44	0.255702E-02		
21	0.9918		-0.0002	0.0006	-0.40	-0.268955E-03		
22	1.0078		-0.0017	0.0006	-2.73	-0.177146E-02		
23	0.9385		0.0000	0.0003	-0.14	-0.458826E-04		
24	0.9926		0.0007	0.0006	1.08	0.700236E-03		
25	1.0009		-0.0007	0.0007	-1.04	-0.751334E-03		
26	0.9987		-0.0032	0.0015	-2.11	-0.321291E-02		
27	1.0063		-0.0004	0.0007	-0.56	-0.442369E-03		
28	0.9980		0.0120	0.0017	6.93	0.120326E-01		
29	0.9865		-0.0023	0.0006	-3.33	-0.231117E-02		
30	0.9987		-0.0078	0.0023	-3.41	-0.789921E-02		
31	1.0013		0.0019	0.0012	1.49	0.190849E-02		
32	1.0018		0.0014	0.0012	1.17	0.141855E-02		
33	0.9934		-0.0029	0.0007	-3.89	-0.293936E-02		
34	0.9622		-0.0010	0.0003	-2.70	-0.102036E-02		
35	1.0055		0.0021	0.0007	2.99	0.218696E-02		
36	-0.2991		-0.0017	0.0000	-43.96	-0.176538E-02		
37	1.0156		-0.0008	0.0004	-2.14	-0.862212E-03		
38	0.9381		0.0000	0.0002	-0.04	-0.112796E-04		
39	0.9747		-0.0000	0.0002	-0.17	-0.529188E-04		
40	0.9962		0.0004	0.0009	0.42	0.413589E-03		
41	0.9527		0.0004	0.0003	1.40	0.472265E-03		
42	0.9914		0.0012	0.0005	2.18	0.129391E-02		
43	0.9972		0.0011	0.0014	0.79	0.114000E-02		
44	1.0006		0.0009	0.0039	0.23	0.912335E-03		
45	0.9989		0.0014	0.0020	0.71	0.147520E-02		
46	1.0014		-0.0013	0.0012	-1.04	-0.136015E-02		
	CONSTANT		MULT F	DF1	DF2	RSQR	RESIDUAL	
	0.271278E-01		24.57	44	705	0.605	0.1325987E-05	

EXHIBIT 3

MODEL FOR VOID USING INTERACTIONS

X	RSQR	X	B	COEF	SE (B)	T	ANALYSIS OF Y
1	0.9995		0.0043		0.0024	1.76	0.431996E-02
2	0.9985		0.0021		0.0011	1.86	0.211242E-02
3	0.9997		0.0023		0.0045	0.51	0.236535E-02
4	0.9999		0.0061		0.0051	1.18	0.616249E-02
5	0.9994		0.0028		0.0017	1.58	0.284498E-02
6	0.9999		-0.0142		0.0056	-2.51	-0.142210E-01
7	1.0008		0.0075		0.0030	2.50	0.751034E-02
8	0.9993		-0.0042		0.0024	-1.76	-0.426057E-02
9	0.9899		0.0010		0.0015	0.65	0.102813E-02
11	0.9998		-0.0045		0.0070	-0.64	-0.450552E-02
12	0.9999		-0.0060		0.0054	-1.11	-0.601794E-02
14	0.9999		0.0129		0.0054	2.40	0.129951E-01
15	0.9997		-0.0154		0.0051	-2.99	-0.154519E-01
16	0.9979		0.0043		0.0014	2.98	0.436891E-02
17	0.9988		-0.0009		0.0028	-0.34	-0.995331E-03
18	0.9984		0.0018		0.0016	1.11	0.181176E-02
19	0.9953		-0.0023		0.0008	-2.60	-0.230708E-02
20	1.0121		0.0025		0.0004	5.44	0.255993E-02
21	0.9918		-0.0003		0.0006	-0.51	-0.338032E-03
22	1.0078		-0.0015		0.0006	-2.46	-0.159503E-02
23	0.9385		-0.0000		0.0003	-0.23	-0.731911E-04
24	0.9926		0.0005		0.0006	0.83	0.542317E-03
25	1.0090		-0.0004		0.0007	-0.56	-0.407382E-03
26	0.9987		-0.0034		0.0015	-2.26	-0.344462E-02
27	1.0063		-0.0004		0.0007	-0.54	-0.425478E-03
28	0.9980		0.0126		0.0017	7.29	0.126646E-01
29	0.9865		-0.0023		0.0006	-3.42	-0.237682E-02
30	0.9987		-0.0084		0.0023	-3.64	-0.844533E-02
31	1.0013		0.0019		0.0012	1.53	0.194862E-02
32	1.0018		0.0014		0.0012	1.23	0.149143E-02
33	0.9934		-0.0031		0.0007	-4.12	-0.311846E-02
34	0.9622		-0.0010		0.0003	-2.68	-0.100988E-02
35	1.0055		0.0022		0.0007	3.14	0.229529E-02
36	-0.2991		-0.0017		0.0000	-44.27	-0.177900E-02
37	1.0156		-0.0010		0.0004	-2.58	-0.104268E-02
38	0.9381		0.0000		0.0002	0.03	0.803825E-05
39	0.9747		-0.0001		0.0002	-0.43	-0.129724E-03
40	0.9962		0.0005		0.0009	0.60	0.595277E-03
41	0.9527		0.0004		0.0003	1.38	0.454385E-03
42	0.9914		0.0012		0.0005	2.17	0.129147E-02
43	0.9972		0.0011		0.0014	0.81	0.116360E-02
44	1.0006		0.0006		0.0039	0.16	0.648976E-03
45	0.9989		0.0015		0.0020	0.73	0.153012E-02
46	1.0014		-0.0013		0.0012	-1.05	-0.136809E-02
CONSTANT		MULT F	DF1	DF2	RSQR	RESIDUAL	
0.210579E-01		24.76	44	705	0.607	0.1327772E-05	

EXHIBIT 4

MODEL FOR STRENGTH USING INTERACTIONS

X	RSQR	X	B	COEF	SE(B)	T
1	0.9995		-711.	8366	147.1942	-4.83
2	0.9985		-270.	0558	68.2929	-3.95
3	0.9997		-332.	0055	275.6882	-1.20
4	0.9999		-64.	5635	312.2766	-0.20
5	0.9994		-1092.	5688	108.2423	-10.09
6	0.9999		2241.	1455	340.7495	6.57
7	1.0008		-2073.	5190	180.9300	-11.46
8	0.9993		418.	9835	145.5554	2.87
9	0.9899		-408.	9157	94.8270	-4.31
11	0.9998		1067.	5815	423.6339	2.52
12	0.9999		508.	3200	326.0465	1.55
14	0.9999		-2045.	2849	326.2623	-6.26
15	0.9997		2771.	8266	311.4108	8.90
16	0.9979		-384.	6616	88.3513	-4.35
17	0.9988		-149.	4142	172.1502	-0.86
18	0.9984		-64.	6407	97.8177	-0.66
19	0.9953		168.	1247	53.4379	3.14
20	1.0121		-46.	4051	28.3200	-1.63
21	0.9918		214.	9610	39.7407	5.40
22	1.0078		-335.	6113	39.0048	-8.60
23	0.9385		38.	3354	18.8933	2.02
24	0.9926		292.	9636	38.9793	7.51
25	1.0090		-758.	8901	43.4694	-17.45
26	0.9987		837.	6492	91.5216	9.15
27	1.0063		-248.	6733	47.2865	-5.25
28	0.9980		-1779.	7412	104.5699	-17.01
29	0.9865		159.	4689	41.8363	3.81
30	0.9987		1336.	7436	139.6617	9.57
31	1.0013		-141.	6998	76.7382	-1.84
32	1.0018		-61.	1337	73.0579	-0.83
33	0.9934		595.	6982	45.5151	13.08
34	0.9622		61.	9687	22.7101	2.72
35	1.0055		-573.	8228	44.0088	-13.03
36	-0.2991		173.	8724	2.4218	71.79
37	1.0156		401.	2891	24.2828	16.52
38	0.9381		-49.	4943	14.8462	-3.33
39	0.9747		-11.	0186	17.8277	-0.61
40	0.9962		-316.	4359	59.2106	-5.34
41	0.9527		3.	1091	20.2654	0.15
42	0.9914		-64.	7191	35.7152	-1.81
43	0.9972		38.	1241	86.0141	0.44
44	1.0006		809.	8768	238.0842	3.40
45	0.9989		-278.	9562	125.1175	-2.22
46	1.0014		96.	0971	78.1869	1.22
	CONSTANT		MULT F	DF1	DF2	RESIDUAL
	561.53027		91.16	44	705	0.850 4822.89942

**EXHIBIT 5A**  
**COMPUTER PROGRAM FOR GENERATING**  
**PREDICTION LEVELS FOR VARIABILITY STUDY**

```

// * PROG. TO FILE IN FFFF2 A PRED SET - A GIVEN SET OF LEVELS AND AN
// * INCREMENT OVER - ORIG AND UNDER AND LEAVE AT ORIG PRINT
// * THEN PROCEED TO NEXT VARIABLE. DATSN 9 FOR PRINT
// * LIST SOURCE PROGRAM
* 1000 4000 INTESRS
* 1005 4000 PRINTN, DISK
* 1010 4000 FILE(4/751160.0,U,IFA)
* 1015 4000 DD(01) 200
* 1020 4000 DD(01) 200
* 1025 4000 IF(102)=D(43)+44
* 1030 4000 I02=D(02)+X02*I02
* 1035 4000 GO(TO) 200
* 1040 4000 103=D(02)
* 1045 4000 103=D(03)+15+5+6
* 1050 4000 DD(03)=D(03)-X03*I03
* 1055 4000 GO(TO) 200
* 1060 4000 104=D(04)+7,7,8
* 1065 4000 DD(04)=D(04)-X04*I04
* 1070 4000 GO(TO) 200
* 1075 4000 DD(04)=D(04)
* 1080 4000 105=D(05)+9+9+10
* 1085 4000 DD(05)=D(05)-X05*I05
* 1090 4000 GO(TO) 200
* 1095 4000 106=D(05)+45+46
* 1100 4000 DD(06)=D(06)-X06*I06
* 1105 4000 GO(TO) 200
* 1110 4000 107=D(06)
* 1115 4000 107=D(07)+11,11,12
* 1120 4000 DD(07)=D(07)-X07*I07
* 1125 4000 GO(TO) 200
* 1130 4000 108=D(07)
* 1135 4000 108=D(08)+13,13+14
* 1140 4000 DD(08)=D(08)-X08*I08
* 1145 4000 GO(TO) 200
* 1150 4000 109=D(09)+15,15+16
* 1155 4000 DD(09)=D(09)-X09*I09
* 1160 4000 GO(TO) 200
* 1165 4000 DD(09)=D(09)
* 1170 4000 GO(TO) 340
* 1175 4000 NORS=JOBS+19,KK1
* 1180 4000 CALL DATSW(19,KK1)
* 1185 4000 IF(108)=200,NORS
* 1190 4000 WRITE(31400,NORS)
* 1195 4000 35000 FORMAT(3137),DD(110F12.3)
* 1200 4000 CONTINUE(NORS)
* 1205 4000 WRITE(4,NORS)
* 1210 4000 CHANGE(CARD FOR NO OF PREDs - IN POS. 12
* 1215 4000 340
* 1220 4000 40,410,410
* 1225 4000 GO(TO) 340
* 1230 4000 101=JOBS+1
* 1235 4000 DD(1111111111)
* 1240 4000 WRITE(4,1111111111)
* 1245 4000 35000 FORMAT(3149),JOBS
* 1250 4000 WRITE(3149,JOBS)
* 1255 4000 49
* 1260 4000 PAUSE(3333)
* 1265 4000 CALL EXIT
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EXHIBIT 5B  
MATRIX OF PREDICTION LEVELS  
FOR VARIABILITY STUDY EXPRESSED IN ENGINEERING UNITS

**EXHIBIT 5C**  
**PROGRAM TO PRODUCE INTERACTIONS**  
**AND CODE FOR GARDNER'S REGRESSION PROGRAM**

EXHIBIT 5D

#### UNREFERENCED STATEMENTS

**EXHIBIT 5E**  
**RESULTS OF VARIABILITY ANALYSIS**

ARIABLES	SETTING AMTS.	STRENGTH	THICKNESS	VOID
Hot Zone Temp.	751 1860°F +15 752 -15	579 600 620 640 660 680 700 720 740 760 780 800 820 840 860 880 900 920 940 960 980 1000 1020 1040 1060 1080 1100 1120 1140 1160 1180 1200 1220 1240 1260 1280 1300 1320 1340 1360 1380 1400 1420 1440 1460 1480 1500 1520 1540 1560 1580 1600 1620 1640 1660 1680 1700 1720 1740 1760°F +15 1780 -15	3.0 E (-1) 3.03560 3.06677 3.09623 3.12323 3.15323 3.18323 3.21323 3.24323 3.27323 3.30323 3.33323 3.36323 3.39323 3.42323 3.45323 3.48323 3.51323 3.54323 3.57323 3.60323 3.63323 3.66323 3.69323 3.72323 3.75323 3.78323 3.81323 3.84323 3.87323 3.90323 3.93323 3.96323 3.99323 4.02323 4.05323 4.08323 4.11323 4.14323 4.17323 4.20323 4.23323 4.26323 4.29323 4.32323 4.35323 4.38323 4.41323 4.44323 4.47323 4.50323 4.53323 4.56323 4.59323 4.62323 4.65323 4.68323 4.71323 4.74323 4.77323 4.80323 4.83323 4.86323 4.89323 4.92323 4.95323 4.98323 5.01323 5.04323 5.07323 5.10323 5.13323 5.16323 5.19323 5.22323 5.25323 5.28323 5.31323 5.34323 5.37323 5.40323 5.43323 5.46323 5.49323 5.52323 5.55323 5.58323 5.61323 5.64323 5.67323 5.70323 5.73323 5.76323 5.79323 5.82323 5.85323 5.88323 5.91323 5.94323 5.97323 6.00323 6.03323 6.06323 6.09323 6.12323 6.15323 6.18323 6.21323 6.24323 6.27323 6.30323 6.33323 6.36323 6.39323 6.42323 6.45323 6.48323 6.51323 6.54323 6.57323 6.60323 6.63323 6.66323 6.69323 6.72323 6.75323 6.78323 6.81323 6.84323 6.87323 6.90323 6.93323 6.96323 6.99323 7.02323 7.05323 7.08323 7.11323 7.14323 7.17323 7.20323 7.23323 7.26323 7.29323 7.32323 7.35323 7.38323 7.41323 7.44323 7.47323 7.50323 7.53323 7.56323 7.59323 7.62323 7.65323 7.68323 7.71323 7.74323 7.77323 7.80323 7.83323 7.86323 7.89323 7.92323 7.95323 7.98323 8.01323 8.04323 8.07323 8.10323 8.13323 8.16323 8.19323 8.22323 8.25323 8.28323 8.31323 8.34323 8.37323 8.40323 8.43323 8.46323 8.49323 8.52323 8.55323 8.58323 8.61323 8.64323 8.67323 8.70323 8.73323 8.76323 8.79323 8.82323 8.85323 8.88323 8.91323 8.94323 8.97323 9.00323 9.03323 9.06323 9.09323 9.12323 9.15323 9.18323 9.21323 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12.69323 12.72323 12.75323 12.78323 12.81323 12.84323 12.87323 12.90323 12.93323 12.96323 13.00323 13.03323 13.06323 13.09323 13.12323 13.15323 13.18323 13.21323 13.24323 13.27323 13.30323 13.33323 13.36323 13.39323 13.42323 13.45323 13.48323 13.51323 13.54323 13.57323 13.60323 13.63323 13.66323 13.69323 13.72323 13.75323 13.78323 13.81323 13.84323 13.87323 13.90323 13.93323 13.96323 14.00323 14.03323 14.06323 14.09323 14.12323 14.15323 14.18323 14.21323 14.24323 14.27323 14.30323 14.33323 14.36323 14.39323 14.42323 14.45323 14.48323 14.51323 14.54323 14.57323 14.60323 14.63323 14.66323 14.69323 14.72323 14.75323 14.78323 14.81323 14.84323 14.87323 14.90323 14.93323 14.96323 15.00323 15.03323 15.06323 15.09323 15.12323 15.15323 15.18323 15.21323 15.24323 15.27323 15.30323 15.33323 15.36323 15.39323 15.42323 15.45323 15.48323 15.51323 15.54323 15.57323 15.60323 15.63323 15.66323 15.69323 15.72323 15.75323 15.78323 15.81323 15.84323 15.87323 15.90323 15.93323 15.96323 16.00323 16.03323 16.06323 16.09323 16.12323 16.15323 16.18323 16.21323 16.24323 16.27323 16.30323 16.33323 16.36323 16.39323 16.42323 16.45323 16.48323 16.51323 16.54323 16.57323 16.60323 16.63323 16.66323 16.69323 16.72323 16.75323 16.78323 16.81323 16.84323 16.87323 16.90323 16.93323 16.96323 17.00323 17.03323 17.06323 17.09323 17.12323 17.15323 17.18323 17.21323 17.24323 17.27323 17.30323 17.33323 17.36323 17.39323 17.42323 17.45323 17.48323 17.51323 17.54323 17.57323 17.60323 17.63323 17.66323 17.69323 17.72323 17.75323 17.78323 17.81323 17.84323 17.87323 17.90323 17.93323 17.96323 18.00323 18.03323 18.06323 18.09323 18.12323 18.15323 18.18323 18.21323 18.24323 18.27323 18.30323 18.33323 18.36323 18.39323 18.42323 18.45323 18.48323 18.51323 18.54323 18.57323 18.60323 18.63323 18.66323 18.69323 18.72323 18.75323 18.78323 18.81323 18.84323 18.87323 18.90323 18.93323 18.96323 19.00323 19.03323 19.06323 19.09323 19.12323 19.15323 19.18323 19.21323 19.24323 19.27323 19.30323 19.33323 19.36323 19.39323 19.42323 19.45323 19.48323 19.51323 19.54323 19.57323 19.60323 19.63323 19.66323 19.69323 19.72323 19.75323 19.78323 19.81323 19.84323 19.87323 19.90323 19.93323 19.96323 20.00323 20.03323 20.06323 20.09323 20.12323 20.15323 20.18323 20.21323 20.24323 20.27323 20.30323 20.33323 20.36323 20.39323 20.42323 20.45323 20.48323 20.51323 20.54323 20.57323 20.60323 20.63323 20.66323 20.69323 20.72323 20.75323 20.78323 20.81323 20.84323 20.87323 20.90323 20.93323 20.96323 21.00323 21.03323 21.06323 21.09323 21.12323 21.15323 21.18323 21.21323 21.24323 21.27323 21.30323 21.33323 21.36323 21.39323 21.42323 21.45323 21.48323 21.51323 21.54323 21.57323 21.60323 21.63323 21.66323 21.69323 21.72323 21.75323 21.78323 21.81323 21.84323 21.87323 21.90323 21.93323 21.96323 22.00323 22.03323 22.06323 22.09323 22.12323 22.15323 22.18323 22.21323 22.24323 22.27323 22.30323 22.33323 22.36323 22.39323 22.42323 22.45323 22.48323 22.51323 22.54323 22.57323 22.60323 22.63323 22.66323 22.69323 22.72323 22.75323 22.78323 22.81323 22.84323 22.87323 22.90323 22.93323 22.96323 23.00323 23.03323 23.06323 23.09323 23.12323 23.15323 23.18323 23.21323 23.24323 23.27323 23.30323 23.33323 23.36323 23.39323 23.42323 23.45323 23.48323 23.51323 23.54323 23.57323 23.60323 23.63323 23.66323 23.69323 23.72323 23.75323 23.78323 23.81323 23.84323 23.87323 23.90323 23.93323 23.96323 24.00323 24.03323 24.06323 24.09323 24.12323 24.15323 24.18323 24.21323 24.24323 24.27323 24.30323 24.33323 24.36323 24.39323 24.42323 24.45323 24.48323 24.51323 24.54323 24.57323 24.60323 24.63323 24.66323 24.69323 24.72323 24.75323 24.78323 24.81323 24.84323 24.87323 24.90323 24.93323 24.96323 25.00323 25.03323 25.06323 25.09323 25.12323 25.15323 25.18323 25.21323 25.24323 25.27323 25.30323 25.33323 25.36323 25.39323 25.42323 25.45323 25.48323 25.51323 25.54323 25.57323 25.60323 25.63323 25.66323 25.69323 25.72323 25.75323 25.78323 25.81323 25.84323 25.87323 25.90323 25.93323 25.96323 26.00323 26.03323 26.06323 26.09323 26.12323 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29.51323 29.54323 29.57323 29.60323 29.63323 29.66323 29.69323 29.72323 29.75323 29.78323 29.81323 29.84323 29.87323 29.90323 29.93323 29.96323 30.00323 30.03323 30.06323 30.09323 30.12323 30.15323 30.18323 30.21323 30.24323 30.27323 30.30323 30.33323 30.36323 30.39323 30.42323 30.45323 30.48323 30.51323 30.54323 30.57323 30.60323 30.63323 30.66323 30.69323 30.72323 30.75323 30.78323 30.81323 30.84323 30.87323 30.90323 30.93323 30.96323 31.00323 31.03323 31.06323 31.09323 31.12323 31.15323 31.18323 31.21323 31.24323 31.27323 31.30323 31.33323 31.36323 31.39323 31.42323 31.45323 31.48323 31.51323 31.54323 31.57323 31.60323 31.63323 31.66323 31.69323 31.72323 31.75323 31.78323 31.81323 31.84323 31.87323 31.90323 31.93323 31.96323 32.00323 32.03323 32.06323 32.09323 32.12323 32.15323 32.18323 32.21323 32.24323 32.27323 32.30323 32.33323 32.36323 32.39323 32.42323 32.45323 32.48323 32.51323 32.54323 32.57323 32.60323 32.63323 32.66323 32.69323 32.72323 32.75323 32.78323 32.81323 32.84323 32.87323 32.90323 32.93323 32.96323 33.00323 33.03323 33.06323 33.09323 33.12323 33.15323 33.18323 33.21323 33.24323 33.27323 33.30323 33.33323 33.36323 33.39323 33.42323 33.45323 33.48323 33.51323 33.54323 33.57323 33.60323 33.63323 33.66323 33.69323 33.72323 33.75323 33.78323 33.81323 33.84323 33.87323 33.90323 33.93323 33.96323 34.00323 34.03323 34.06323 34.09323 34.12323 34.15323 34.18323 34.21323 34.24323 34.27323 34.30323 34.33323 34.36323 34.39323 34.42323 34.45323 34.48323 34.51323 34.54323 34.57323 34.60323 34.63323 34.66323 34.69323 34.72323 34.75323 34.78323 34.81323 34.84323 34.87323 34.90323 34.93323 34.96323 35.00323 35.03323 35.06323 35.09323 35.12323 35.15323 35.18323 35.21323 35.24323 35.27323 35.30323 35.33323 35.36323 35.39323 35.42323 35	

GODDARD SPACE FLIGHT CENTER  
BUSINESS DATA BRANCH  
ADDRESS LABEL SYSTEM  
RUN DATE - SEP-01-1970

MASTER LIST  
REPORT NO 1932  
LIST 020

✓ WINNIE M. MORGAN  
NASA HEADQUARTERS  
CODE US 2

✓ 1 E. M. COHN  
2 NASA HEADQUARTERS  
3 CODE RNW

✓ 0001 *label #*  
GERALD HALPERT  
✓ GSFC  
CODE 764

✓ 1 7 8 0003  
2 3 THOMAS HENNIGAN  
3 4 GSFC  
4 5 CODE 761  
5 6  
6 7  
7 8 0007

✓ 0006  
JOHN L. PATTERSON  
MS 472  
LANGLEY RESEARCH CENTER

✓ 1 3 7 8 0011  
2 3 M. B. SEYFFERT  
3 4 MS 112  
4 5 LANGLEY RESEARCH CENTER  
5 6  
6 7  
7 8 0016

✓ W. E. RICE  
MANNED SPACECRAFT CENTER  
CODE EP-5

✓ 1 3 7 8 0016  
2 3 JON RUBENZER  
3 4 BIOSATELLITE PROJECT  
4 5 AMES RESEARCH CENTER  
5 6 CODE PBS, M. S. 244-2  
6 7  
7 8 0016

✓ 0015

✓ DR. R. LUTWACK, MS 198-220  
JET PROPULSION LABORATORY  
4800 OAK GROVE DRIVE  
PASADENA, CAL 91103

✓ 1 3 7 8 0020  
2 3 MR. ALVIN UCHIYAMA, MS 198-223  
3 4 JET PROPULSION LABORATORY  
4 5 4800 OAK GROVE DRIVE  
5 6 PASADENA, CAL 91103  
6 7  
7 8 0020

✓ 0019

✓ L. A. SPANO  
U.S. ARMY NATICK LAB.  
CLOTHING & ORGANIC MATERIALS  
DIV.  
NATICK, MASSACHUSETTS 01760

✓ 1 3 7 8 0026  
2 3 NATHAN KAPLAN  
2 4 HARRY DIAMOND LABORATORIES  
3 5 ROOM 300, BUILDING 92  
4 6 CONN. AVE. & VAN NESS ST., N.W.  
5 7 WASHINGTON, D. C. 20438  
6 8 0026

✓ 0025

✓ DR. J. C. WHITE  
NAVAL RESEARCH LABORATORY  
WASHINGTON, D. C. 20390  
CODE 6160

✓ 1 3 7 8 0031  
2 3 NAVAL SHIP R&D CENTER  
3 4 ATTN. J. H. HARRISON  
4 5 CODE M760  
5 6 ANNAPOLIS, MARYLAND 21402  
6 7  
7 8 0031

✓ 0030

✓ PHILIP B. COLE  
NAVAL ORDNANCE LABORATORY  
SILVER SPRING, MARYLAND 20910  
CODE 232

✓ 1 3 7 8 0035  
2 3 NAVAL SHIP ENGINEERING CENTER  
3 4 ATTN. C. F. VIGLIOTTI  
4 5 6157D  
5 6 WASHINGTON, D. C. 20360  
6 7  
7 8 0035

✓ 0034

*sf#*

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1  
2 A. M. GREG ANDRUS  
3 NASA HEADQUARTERS  
4 CODE SAC  
5  
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7 0004 8 0005

1  
2 MR. CHARLES B. GRAFF  
3 S&E-ASTR-EP  
4 ✓ GEORGE MARSHALL SFC  
5 NATIONAL AERO. & SPACE ADMIN.  
6 HUNTSVILLE, ALABAMA 35812  
7  
8

1  
2 JOSEPH SHERFEEY  
3 GSFC  
4 CODE 764  
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8

0008 8 0009

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2 DR. LOUIS ROSENBLUM  
3 ✓ LEWIS RESEARCH CENTER  
4 STOP 302-1  
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2 DR. J. STEWART FORDYCE  
3 LEWIS RESEARCH CENTER  
4 STOP 6-1  
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0012 8 0014

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2 DR. SOL GILMAN  
3 ✓ ELECTRONICS RESEARCH CENTER  
4 CODE CPE  
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2 MR. PAUL GOLDSMITH  
3 JPL  
4 M.S. 198-223  
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0017 8 0018

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2 ✓ U. S. ARMY  
3 ELECTRO TECHNOLOGY LABORATORY  
4 ENERGY CONVERSION RESEARCH DIV  
5 MERDC  
6 FORT BELVOIR, VIRGINIA 22060  
7  
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2 U. S. ARMY ELECTRONIC R&D. LAB  
3 ATTN. CODE AMSEL-KL-P  
4 FORT MONMOUTH, NEW JERSEY 07703  
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0022 8 0023

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2 ✓ OFFICE OF NAVAL RESEARCH  
3 ATTN. DIR., POWER PROGRAM  
4 CODE 473  
5 WASHINGTON, D. C. 20360  
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2 OFFICE OF NAVAL RESEARCH  
3 ATTN. MR. HARRY FOX  
4 CODE 472  
5 WASHINGTON, D. C. 20360  
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0028 8 0029

1  
2 ✓ MILTON KNIGHT /CODE AIR-340C/  
3 NAVAL WEAPONS CENTER  
4 ✓ NAVAL AIR SYSTEMS COMMAND  
5 DEPARTMENT OF THE NAVY  
6 WASHINGTON, D. C. 20360  
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2 ATTN. WILLIAM C. SPINDLER  
3 CORONA LABORATORIES  
4 CORONA, CALIFORNIA 91720  
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0032 8 0033

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2 ✓ U. S. NAVAL OBSERVATORY  
3 ATTN. ROBERT E. TRUMBLE  
4 STIC, BLDG. 52  
5 WASHINGTON, D. C. 20390  
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2 COMMANDER,  
3 NAVAL SHIP SYSTEMS COMMAND  
4 ✓ ATTN. BERNARD B. ROSENBAUM  
5 /CODE 03422/  
6 DEPARTMENT OF THE NAVY  
7 WASHINGTON, D. C. 20360  
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9

0036 8 0037

GODDARD SPACE FLIGHT CENTER  
BUSINESS DATA BRANCH  
ADDRESS LABEL SYSTEM  
RUN DATE SEP-01-1970

MASTER LIST  
REPORT NO 1932  
LIST 020

AERO PROPULSION LABORATORY  
ATTN. JAMES E. COOPER  
APIP-2  
WRIGHT-PATTERSON AFB, OH. 45433

1 FRANCIS X. DOHERTY  
2 EDWARD RASKIND /WING F/  
3 A F CAMBRIDGE RESEARCH LAB.  
4 ATTN. CRE  
5 L. G. HANSOM FIELD  
6 BEDFORD, MASS 01741

0039

8 0040

A. M. F.  
✓ MR. R. A. KNIGHT  
AMF INC.  
689 HOPE STREET  
STAMFORD, CONN. 06907

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3 LIB. ACQUISITION GROUP  
4 AEROSPACE CORPORATION  
5 P. O. BOX 95085  
6 LOS ANGELES, CAL 90045

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8 0046

DR. C. L. FAUST  
BATTELLE MEMORIAL INSTITUTE  
505 KING AVENUE  
COLUMBUS, OHIO 43201

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2  
3 BELLCOMM  
4 ATTN. B. W. MOSS  
5 1100 17TH STREET, N. W.  
6 WASHINGTON, D. C. 20036

0049

8 0050

DR. HOWARD J. STRAUSS  
BURGESS BATTERY COMPANY  
FOOT OF EXCHANGE STREET  
FREEPORT, ILLINOIS 61033

1  
2  
3 DR. EUGENE WILLIHNANGANZ  
4 C & D BATTERIES  
5 DIV. OF ELECTRIC AUTOLITE CO.  
6 CONSHOHOCKEN, PA. 19428

0053

8 0054

DR. L. J. MINNICK  
G. & W. H. CORSON, INC.  
✓ PLYMOUTH MEETING  
PENNSYLVANIA 19462

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3 CUBIC CORPORATION  
4 ATTN. LIBRARIAN  
5 9233 BALBOA AVENUE  
6 SAN DIEGO, CAL 92123

0058

8 0059

ESB, INC.  
✓ ATTN. DIRECTOR OF ENGINEERING  
P. O. BOX 11097  
RALEIGH, NORTH CAROLINA 27604

1 DR GALEN FRYINGER  
2 ESB INC  
3 CARL F. NORDBERG RESEARCH CTR.  
4 19 WEST COLLEGE AVENUE  
5 YARDLEY, PA 19067

0062

8 0063

ELECTROMITE CORPORATION  
ATTN. R. H. SPARKS  
2117 SOUTH ANNE STREET  
SANTA ANA, CAL 92704

1  
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3 DR. W. P. CADOGAN  
4 EMHART CORPORATION  
5 BOX 1620  
6 HARTFORD, CONNECTICUT 06102

0066

8 0068

DR. ARTHUR FLEISCHER  
✓ 466 SOUTH CENTER STREET  
ORANGE, NEW JERSEY 07050

1 STEPHAN J GASTON  
2 GRUMMAN AEROSPACE CORP  
3 PLANT 35 P.O.D.  
4 BETHPAGE NEW YORK 11714  
5 BETHPAGE, L.I. NEW YORK 11714

0071

8 0072

1	ROME AIR DEVELOPMENT CTR. ESD ATTN. FRANK J. MCLLURA /EMEAM/ GRIFFIS AFB, NEW YORK 13442	2	DR. W. J. HAMER NATIONAL BUREAU OF STANDARDS WASHINGTON, D. C. 20234
0041		8	0043
1	DR. R. T. FCLEY CHEMISTRY DEPARTMENT AMERICAN UNIVERSITY MASS. & NEBRASKA AVENUE, N.W. WASHINGTON, D. C. 20016	2	DR. H. L. RECHT ATOMICS INTERNATIONAL DIVISION NORTH AMERICAN AVIATION, INC. 8900 DE SOTA AVENUE CANOGA PARK, CAL 91304
0047		8	0048
1	MR. LOU BELOVE SONOTONE CORPORATION COLD SPRING, NEWYORK 10516	2	DR. CARL BERGER 13401 KOOTENAY DRIVE SANTA ANA, CAL 92705
0051		8	0052
1	CALVIN COLLEGE, SCIENCE BLDG. ATTN. PROF. T. P. DIRKSE 3175 BURTON STREET, S. E. GRAND RAPID, MICHIGAN 49506	2	H. GOLDSMITH CATALYST RESEARCH CORPORATION 6101 FALLS ROAD BALTIMORE, MARYLAND 21209
0055		8	0056
1	DELCO REMY DIVISION ATTN. J. A. KERALLA GENERAL MOTORS CORPORAION 2401 COLUMBUS AVENUE ANDERSON, INDIANA 46011	2	E. I. DU PONT NEMOURS & CO. ATTN. J. M. WILLIAMS ENGINEERING MATERIALS LAB EXPERIMENTAL STATION, BLDG. 304 WILMINGTON, DELAWARE 19898
0060		8	0061
1	E. P. BROGLIO EAGLE-PICHER COMPANY P. O. BOX 47 JOPLIN, MISSOURI 64801	2	DR. MORRIS EISENBERG ELECTROCHIMICA CORPORATION 1140 O'BRIEN DRIVE MENLO PARK, CAL 94025
0064		8	0065
1	DR. H. G. OSWIN ENERGETICS SCIENCE, INC. 4461 BRONX BLVD. NEW YORK, NEW YORK 10475	2	MR. D. C. FEDER BELL LABORATORIES MURRAY HILL, NEW JERSEY 07974
0069		8	0070
1	GENERAL ELECTRIC COMPANY ATTN DR W N CARSON RESEARCH & DEVELOPMENT CENTER P. O. BOX 43 SCHNECTADY, NEW YORK 12301	2	GENERAL ELECTRIC COMPANY ATTN MR H THIERFELDER MISSILE & SPACE DIVISION SPACECRAFT DEPARTMENT P. O. BOX 8555 PHILADELPHIA, PA. 19101
0073		8	0074

GODDARD SPACE FLIGHT CENTER  
BUSINESS DATA BRANCH  
ADDRESS LABEL SYSTEM  
RUN DATE SEP-01-1970

MASTER LIST  
REPORT NO 1932  
LIST 020

GENERAL ELECTRIC COMPANY  
ATTN. WHITNEY LIBRARY  
P. O. BOX 8  
SCHENECTADY, NEW YORK 12301

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